



Studies toward the total synthesis of (–)-kampanol A: an efficient construction of the ABCD ring system

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Received 9 August 2002; revised 26 August 2002; accepted 30 August 2002

Abstract—The optically active tetracyclic ABCD ring system **2** of (–)-kampanol A (**1**), a novel Ras farnesyltransferase inhibitor from a microorganism, was efficiently synthesized starting from the known ketol **4** as a model study. The synthetic method involves conjugate addition reaction of the Grignard reagent of the bromobenzene derivative **14** to the α -methylene ketone **10** to form the coupling product **15** and phenylselenium-mediated cyclization reaction of the phenol derivative **17** to stereoselectively construct the requisite tetracyclic intermediate **18** as the pivotal steps. © 2002 Elsevier Science Ltd. All rights reserved.

Kampanol A (**1**), isolated from the culture broth of *Stachybotrys kampalensis* by the Merck research group in 1998, has been shown to be a novel and specific inhibitor of Ras protein farnesyltransferase.^{1,2} This enzyme catalyzes the farnesylation of Ras p21 protein on the cysteine residue of the C-terminal CAAX-tetra-peptide sequence (C: cysteine, A: aliphatic amino acid, X: serine or methionine); this post-translational modification is essential for plasma membrane association that is a critical step in *ras*-mediated oncogenesis.³ Therefore, kampanol A (**1**) is anticipated to be a promising agent for novel cancer therapeutics. The gross structure of **1** including the relative stereochemistry was revealed by extensive spectroscopic studies to have a novel pentacyclic 1*H*-benzo[*a*]furo[3,4-*h*]xanthen-3(6*H*)-one skeleton (ABCDE ring system) with five asymmetric carbons.^{1,4,5} Its remarkable biological properties and unique structural features prompted us to undertake a project directed toward the total synthesis of optically active **1**. In this communication, we wish to disclose our preliminary results concerning an efficient and facile synthetic method for the model compound **2** possessing the tetracyclic ABCD ring system with the requisite substituents and asymmetric carbons contained in **1**. The present study was conducted to explore our synthetic strategy for this fascinating natural product **1**. And furthermore, the model

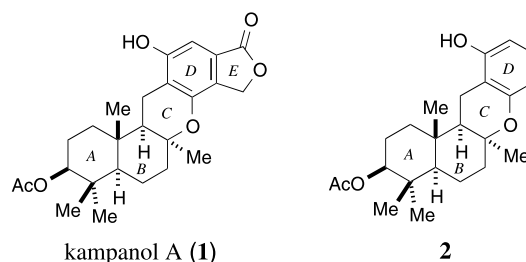


Figure 1. Structures of kampanol A (**1**) and the model compound **2**.

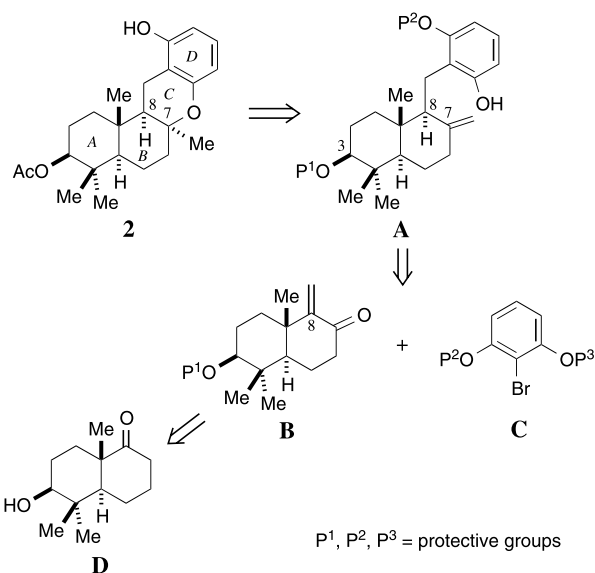
compound **2** will particularly be useful in the structure–activity studies of kampanol A and related compounds. To the best of our knowledge, synthetic studies of **1** have not been reported to date⁶ (Fig. 1).

Our synthetic plan for the model compound **2** is outlined in Scheme 1. We envisioned that the target compound **2** would be elaborated by the stereocontrolled cyclization of the phenol derivative **A** applying the related protocols previously described in the literature^{6a–g} followed by manipulation of the C-3 and the phenolic hydroxy protecting groups. The cyclization precursor **A** would be prepared through the conjugate addition reaction⁷ between the α -methylene ketone **B** and the Grignard reagent of the *ortho*-disubstituted bromobenzene derivative **C**, where we expected that the C-8 substituent in the coupling product should be placed in an equatorial orientation under thermodynamically and/or kinetically controlled reaction conditions.⁸ The key segments **B** and **C**, in turn, would be obtained from the known *trans*-decalone **D**⁹ and the

Keywords: kampanol A; Ras farnesyltransferase inhibitor; conjugate addition; phenylselenium-mediated cyclization.

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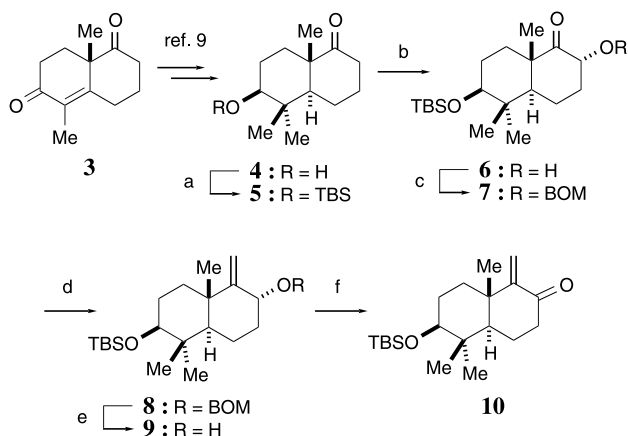
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Scheme 1. Synthetic plan for the model compound **2**.

commercially available resorcinol (**11**, Scheme 3), respectively.

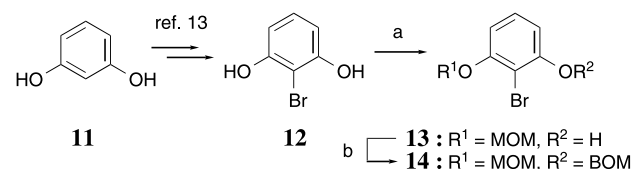
We initially pursued the synthesis of the decalin segment **10** (corresponding to **B**) as shown in Scheme 2. Although the compound **10** has been previously synthesized by Seifert et al.^{8a} starting with the (+)-Wieland Miescher ketone in 21% overall yield in nine steps, we sought an alternative, more efficient and reliable method for the synthesis of **10**. We have now found that **10** can be synthesized starting from (+)-Wieland Miescher ketone analogue **3** in 44% overall yield in a ten-step sequence. Thus, the known ketol **4**⁹ (corresponding to **D**), readily derived from (+)-**3** (>99% ee) in four steps according to the reported method,⁹ was first converted to the α -hydroxy ketone **6** in 73% overall yield via a two-step sequence involving protection of



Scheme 2. Synthesis of the decalin segment **10**. *Reagents and conditions:* (a) TBSOTf, 2,6-lutidine, CH₂Cl₂, rt, 99%; (b) NaN(TMS)₂, THF, -78°C, 2-phenylsulfonyl-3-phenyloxaziridine, -78°C, 74%; (c) BOMCl, *i*-Pr₂NEt, CH₂Cl₂, rt, 98%; (d) Ph₃P⁺CH₃Br⁻, *t*-BuOK, benzene, reflux, 91%; (e) Li, liq. NH₃-THF, 80%; (f) Dess–Martin periodinane, CH₂Cl₂, rt, 98%.

the hydroxy group in **4** as its *t*-butyldimethylsilyl (TBS) ether followed by oxidation of the resulting TBS ether **5** employing a combination of NaN(TMS)₂ and 2-phenylsulfonyl-3-phenyloxaziridine developed by Davis et al.^{10,11} After protection of the hydroxy group in **6** as its benzylloxymethyl (BOM) ether (98%), the resulting BOM ether **7** was then subjected to Wittig methylenation to provide the *exo*-olefin **8** in 91% yield. Removal of the BOM protecting group in **8** under Birch conditions (Li/liq.NH₃/THF) followed by Dess–Martin oxidation¹² of the resulting alcohol **9**, furnished the desired decalin segment **10**, mp 60–61°C, [α]_D²⁰ -39.0° (*c* 1.02, CHCl₃), in 78% yield for the two steps.

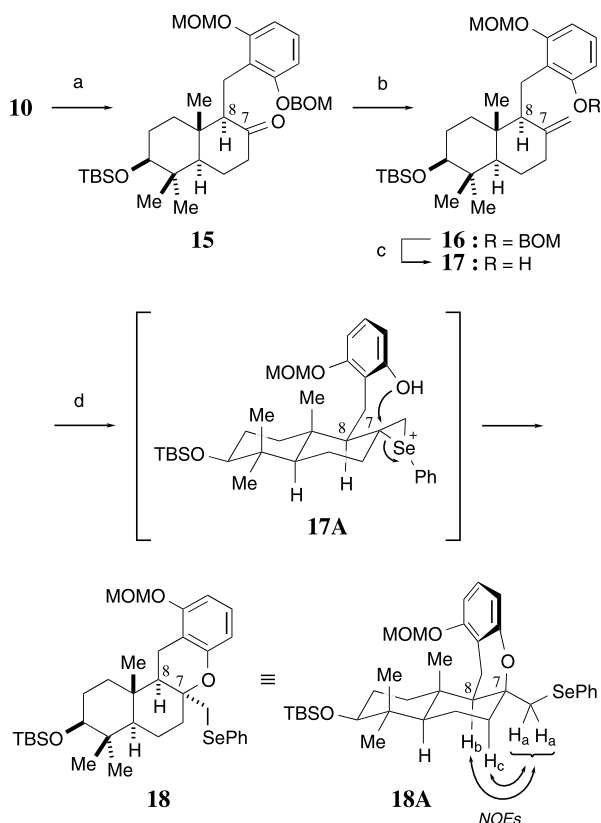
Next, the synthesis of the aromatic segment **14** (corresponding to **C**) was carried out as shown in Scheme 3. The starting material, 2-bromoresorcinol (**12**), was prepared from the commercially available resorcinol (**11**) in two steps according to the literature.¹³ The two hydroxy groups present in **12** were differently protected; that is, monoprotection with a methoxymethyl (MOM) group afforded the MOM ether **13** (41%), which upon further protection as its BOM ether furnished the requisite aromatic segment **14** in 96% yield.



Scheme 3. Synthesis of the aromatic segment **14**. *Reagents and conditions:* (a) MOMCl, *i*-Pr₂NEt, CH₂Cl₂, rt, 41%; (b) BOMCl, *i*-Pr₂NEt, CH₂Cl₂, rt, 96%.

Having obtained both the decalin segment **10** and the aromatic segment **14**, we next focused our attention on the critical coupling reaction of these two segments. As shown in Scheme 4, the conjugate addition of the Grignard reagent, prepared from **14** and Mg turnings in the presence of 1,2-dibromoethane in Et₂O, to the α -methylene ketone **10** proceeded smoothly without the addition of any copper salts,¹⁴ leading to the formation of the desired coupling product **15**, [α]_D²⁰ -26.3° (*c* 1.07, CHCl₃), in 95% yield with complete stereoselectivity at the C-8 position. The coupling product **15** was further converted to the phenol derivative **17** (corresponding to **A**), [α]_D²⁰ -12.6° (*c* 0.79, CHCl₃), the key cyclization precursor, through a two-step sequence of reactions involving Wittig methylenation of the carbonyl function in **15** followed by reductive removal of the BOM protecting group in the resulting *exo*-olefin **16**, [α]_D²⁰ -20.6° (*c* 1.08, CHCl₃), under the Birch conditions (Li/liq.NH₃/THF) (95%).

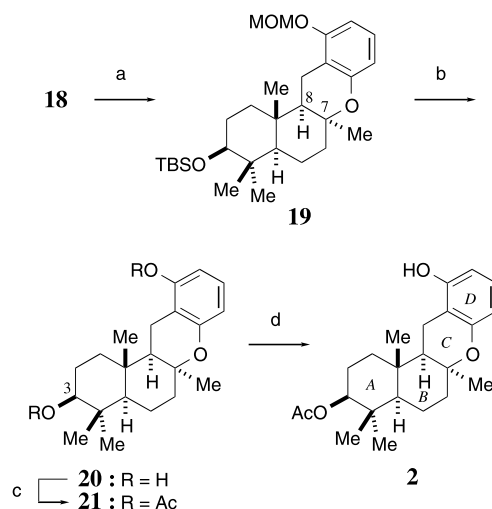
With the key cyclization precursor **17** in hand, our next efforts were directed toward the crucial stereocontrolled cyclization reaction of **17** to construct the requisite tetracyclic ABCD ring system. After several experiments,¹⁵ to our delight, the cyclization reaction of the phenol **17** was successfully achieved by the use of organoselenenylating reagent.¹⁶ Thus, **17** was treated



Scheme 4. Synthesis of the tetracyclic key intermediate **18**. *Reagents and conditions:* (a) **14**, Mg, 1,2-dibromoethane, Et₂O, reflux; **10**, 0°C→rt, 95%; (b) Ph₃P⁺CH₃Br⁻, *t*-BuOK, benzene, reflux, 97%; (c) Li, liq. NH₃-THF, 95%; (d) *N*-phenylselenophthalimide, SnCl₄, CH₂Cl₂, -78°C, 98%.

with *N*-phenylselenophthalimide (1.6 equiv.) in the presence of tin(IV) chloride (1.4 equiv.) in dichloromethane at -78°C for 2 h, resulting in the formation of the desired cyclized product **18**, [α]_D²⁰ +21.5° (*c* 1.04, CHCl₃), as the single isomer in 98% yield, which possesses the correct stereochemistry at the C-7 position.^{6g} This phenylselenium-mediated cyclization reaction would proceed through the transition state such as selenonium ion intermediate **17A**, where the selenonium ion would be opened by the attack of the inner phenolic hydroxy group to provide the 6-*exo* cyclization product **18**. The newly formed stereochemistry at the C-7 position in **18** was proven by NOESY experiments in the 500 MHz ¹H NMR spectrum; thus, as depicted in the formula **18A**, clear NOE interactions between the signals due to H_a and H_b and between the signals due to H_a and H_c were respectively observed.

The final route that led to completion of the synthesis of the target compound **2** is summarized in Scheme 5. Thus, removal of the phenylselenenyl group in **18** by reaction with tri-*n*-butyltin hydride in the presence of 2,2-azobis(isobutyronitrile) (AIBN) followed by complete deprotection of the TBS and MOM groups in the resulting product **19**, [α]_D²⁰ -13.8° (*c* 0.74, CHCl₃), afforded the diol **20**, mp 108–109°C, [α]_D²⁰ -40.6° (*c* 0.91, CHCl₃), in 75% yield for the two steps. Direct conversion of **20** to **2** by selective acetylation of the C-3



Scheme 5. Synthesis of the model compound **2**. *Reagents and conditions:* (a) *n*-Bu₃SnH, AIBN, toluene, reflux, 78%; (b) 6 M HCl, MeOH, 50°C, 96%; (c) Ac₂O, DMAP, pyridine, rt, 85%; (d) *t*-BuOK, THF-*t*-BuOH (5:1), rt, 96%.

hydroxy group met with failure. Therefore, **20** was transformed to **2** via a two-step sequence of reactions; thus, acetylation of both the C-3 and phenolic hydroxy groups in **20** furnished the corresponding diacetate **21** (85%), mp 144.5–146°C, [α]_D²⁰ -8.3° (*c* 1.02, CHCl₃), which upon chemoselective removal of the phenolic acetyl group by exposure to potassium *t*-butoxide (1.05 equiv.) in THF-*t*-butyl alcohol (5:1) at room temperature finally provided **2**,¹⁷ mp 233–234°C, [α]_D²⁰ -10.4° (*c* 0.98, CHCl₃), in 96% yield.

In summary, we have achieved an enantioselective synthesis of the ABCD ring system **2** of (-)-kampanol A (**1**) as a model study. The explored method features a conjugate addition reaction of the Grignard reagent, prepared from the bromobenzene derivative **14**, to the α,β -unsaturated ketone **10** to obtain the coupling product **15** and an organoselenium-mediated cyclization reaction of the phenol derivative **17** to construct the requisite tetracyclic intermediate **18** with complete stereoselectivity. Further investigation toward the total synthesis of kampanol A and its analogues based on this preliminary study, as well as biological evaluation of the model compound **2** are now in progress and will be reported appropriately in due course.

Acknowledgements

This work was supported in part by the Pfizer Award in Synthetic Organic Chemistry, Japan.

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11. The high stereoselectivity observed for the α -hydroxylation of the ketone **5** leading to the α -hydroxyketone **6** can be accounted for by the assumption that the oxidizing reagent approaches from the less hindered α -face of the enolate generated from **5** under the influence of the axial juncture methyl group.
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14. In general, the addition reaction of Grignard reagents to α,β -unsaturated ketones in the absence of copper salts affords 1,2-addition products. However, in this case the 1,4-addition product **15** was only obtained, and this is probably due to severe 1,3-diaxial interactions between the axial juncture methyl group in **10** and the incoming Grignard reagent of **14**.
15. When the phenol derivative **17** was subjected to acid-mediated cyclization reaction (e.g. $BF_3 \cdot Et_2O/CH_2Cl_2/-60^\circ C \rightarrow rt$), the undesired C-7 epimer of **19** was exclusively produced in 86% yield. This stereochemical outcome can be rationalized by considering that the inner phenolic hydroxy group attacks the C-7 tertiary carbocation, in situ generated by acid treatment, from the less hindered α -face of the molecule under the influence of the β -oriented axial methyl group at the decalin junction.
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17. Spectral data for **2**: 1H NMR (500 MHz, $CDCl_3$): δ 6.92 (1H, t, $J=8.0$ Hz), 6.37 (1H, d, $J=8.0$ Hz), 6.30 (1H, dd, $J=0.8, 8.0$ Hz), 4.77 (1H, s), 4.50 (1H, dd, $J=4.7, 11.7$ Hz), 2.72 (1H, d, $J=17.9$ Hz), 2.66 (1H, dd, $J=7.5, 17.9$ Hz), 2.12–2.21 (1H, m), 2.05 (3H, s), 1.91 (1H, dt, $J=3.5, 13.2$ Hz), 1.49–1.77 (5H, m), 1.38 (1H, d, $J=7.5$ Hz), 1.18 (3H, s), 1.09–1.21 (1H, m), 0.97–1.03 (1H, m), 0.90 (3H, s), 0.86 (3H, s), 0.74 (3H, s); ^{13}C NMR (125 MHz, $CDCl_3$): δ 171.4, 155.8, 153.5, 126.6, 109.8, 109.5, 106.2, 81.1, 75.0, 54.4, 48.7, 40.5, 38.0, 37.8, 37.7, 28.4, 26.8, 23.4, 21.3, 17.8, 17.4, 16.8, 14.2; IR (KBr): 3447, 2946, 2361, 1699, 1616, 1595, 1468, 1377, 1277, 1169, 1136, 1084, 1030, 972, 905, 777, 561 cm^{-1} ; EI-MS (m/z): 372 (M^+).